

HW 5.1

Writing formulas for ionic compounds

Name: _____ Band: _____

Write the formulas for the following ionic compounds

1. Lithium chloride _____
2. Calcium oxide _____
3. Magnesium fluoride _____
4. Strontium iodide _____
5. Potassium bromide _____
6. Barium sulfide _____
7. Aluminum phosphide _____
8. Aluminum carbide _____
9. Francium oxide _____
10. Sodium nitride _____
11. Cuprous oxide _____
12. Cupric sulfide _____
13. Stannic sulfide _____
14. Chromous nitride _____
15. Ferric nitride _____
16. Ferrous iodide _____
17. Chromic oxide _____
18. Stannic fluoride _____
19. Stannous chloride _____
20. Plumbic sulfide _____
21. Plumbous nitride _____
22. Manganous iodide _____
23. Cobaltous oxide _____
24. Lithium acetate _____
25. Sodium hydroxide _____
26. Lithium carbonate _____
27. Aluminum nitrate _____
28. Calcium nitrate _____
29. Calcium nitrite _____
30. Barium phosphate _____
31. Potassium chlorate _____
32. Calcium bicarbonate _____
33. Barium sulfate _____
34. Sodium hydrogen carbonate _____
35. Lithium phosphite _____
36. Ammonium carbonate _____
37. Cupric acetate _____
38. Ferrous hydroxide _____