

Lab 1.2
Physical and Chemical Changes

Name: _____
Section: _____

	What to Do:	Make an Observation (Record what you see in this column)	Type of Change Observed: (Physical or Chemical?)
1	Cut Lead into Pieces		
2	Heat Lead (In a test tube)		
3	Cut Paper into Pieces		
4	Heat (Burn) Paper in test tube		
5	Cut Mg into pieces		
6	Place Mg Ribbon into HCl		
7	Dissolve NaCl in H ₂ O		
8	Dissolve NaCl in H ₂ O w/ AgNO ₃		
9	Dissolve sugar in H ₂ O		
10	Dissolve Sugar in H ₂ SO ₄		

Questions: 1.) What Events Illustrate a physical change? _____

2) Why are physical changes important to Chemistry?

3) What events illustrate a chemical change?

4) Why are chemical changes important in chemistry?

5) What happens to heat in a physical change?

6) What happens to heat in a chemical change?
